

CHEMISTRY 
A EUROPEAN JOURNAL

Supporting Information

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**Formation of 1D and 3D coordination polymers in the solid state
induced by mechanochemical and annealing treatments:
bis 3-cyano-pentane-2,4-dionato metal complexes**

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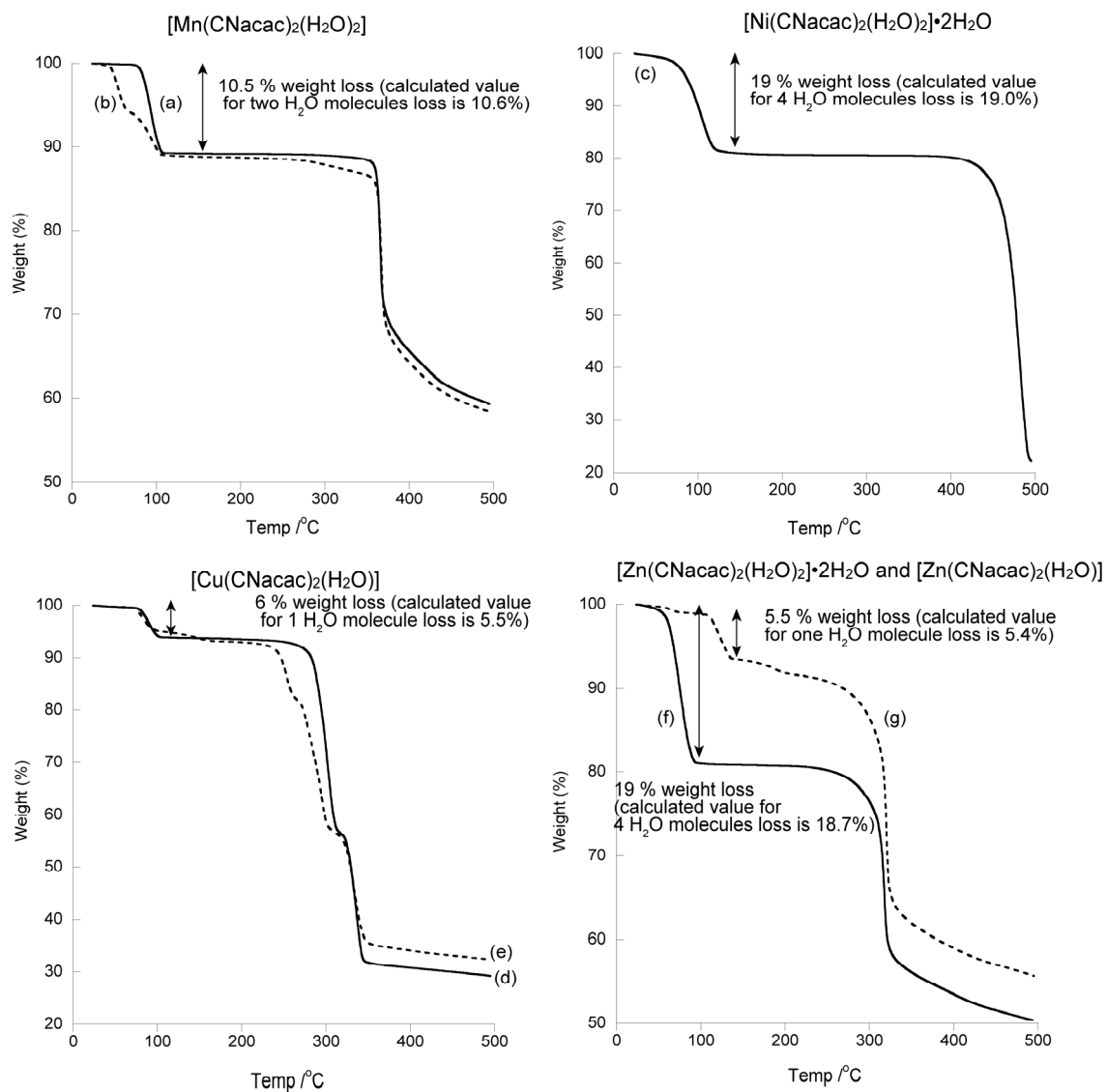


Figure S1. TG curves of (a) [Mn(CNacac)₂(H₂O)₂] (solution reaction), (b) [Mn(CNacac)₂(H₂O)₂] (mechanochemical reaction), (c) [Ni(CNacac)₂(H₂O)₂]·2H₂O (solution), (d) [Cu(CNacac)₂(H₂O)] (solution), (e) [Cu(CNacac)₂(H₂O)] (mechanochemical), (f) [Zn(CNacac)₂(H₂O)₂]·2H₂O (solution) and (g) [Zn(CNacac)₂(H₂O)] (mechanochemical).

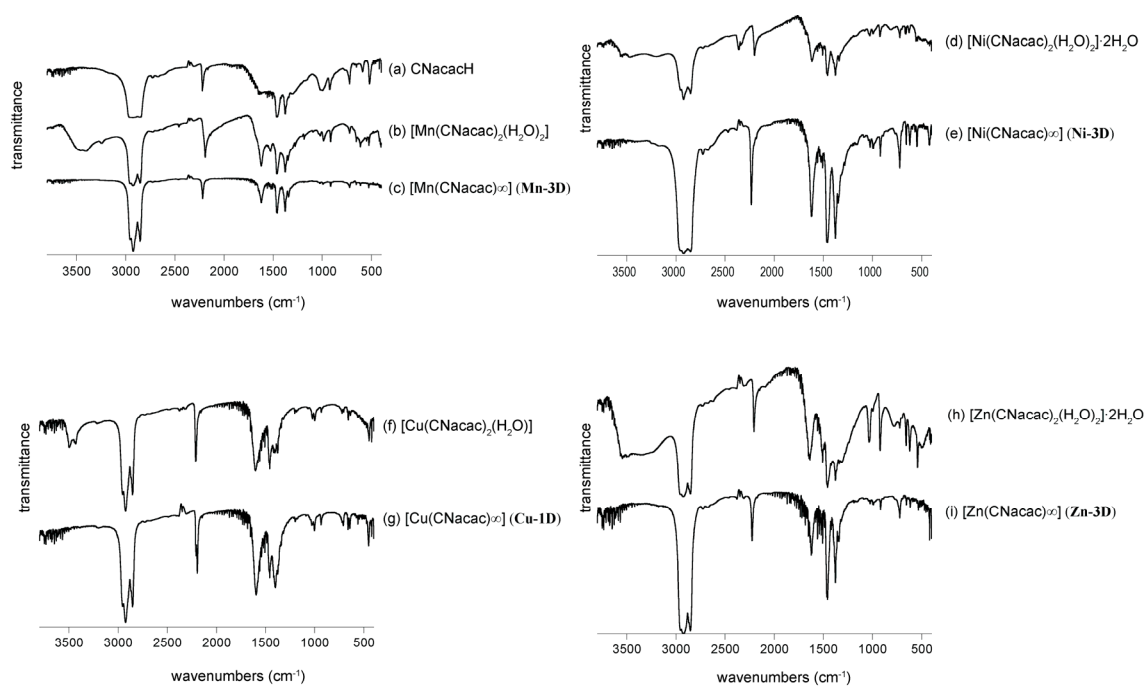


Figure S2. IR spectra of (a) CNaacH, (b) $[\text{Mn}(\text{CNaac})_2(\text{H}_2\text{O})_2]$, (c) **Mn-3D**, (d) $[\text{Ni}(\text{CNaac})_2(\text{H}_2\text{O})_2] \cdot 2\text{H}_2\text{O}$, (e) **Ni-3D**, (f) $[\text{Cu}(\text{CNaac})_2(\text{H}_2\text{O})]$, (g) **Cu-1D**, (h) $[\text{Zn}(\text{CNaac})_2(\text{H}_2\text{O})_2] \cdot 2\text{H}_2\text{O}$ and (i) **Zn-3D**, measured by the Nujol mull method.

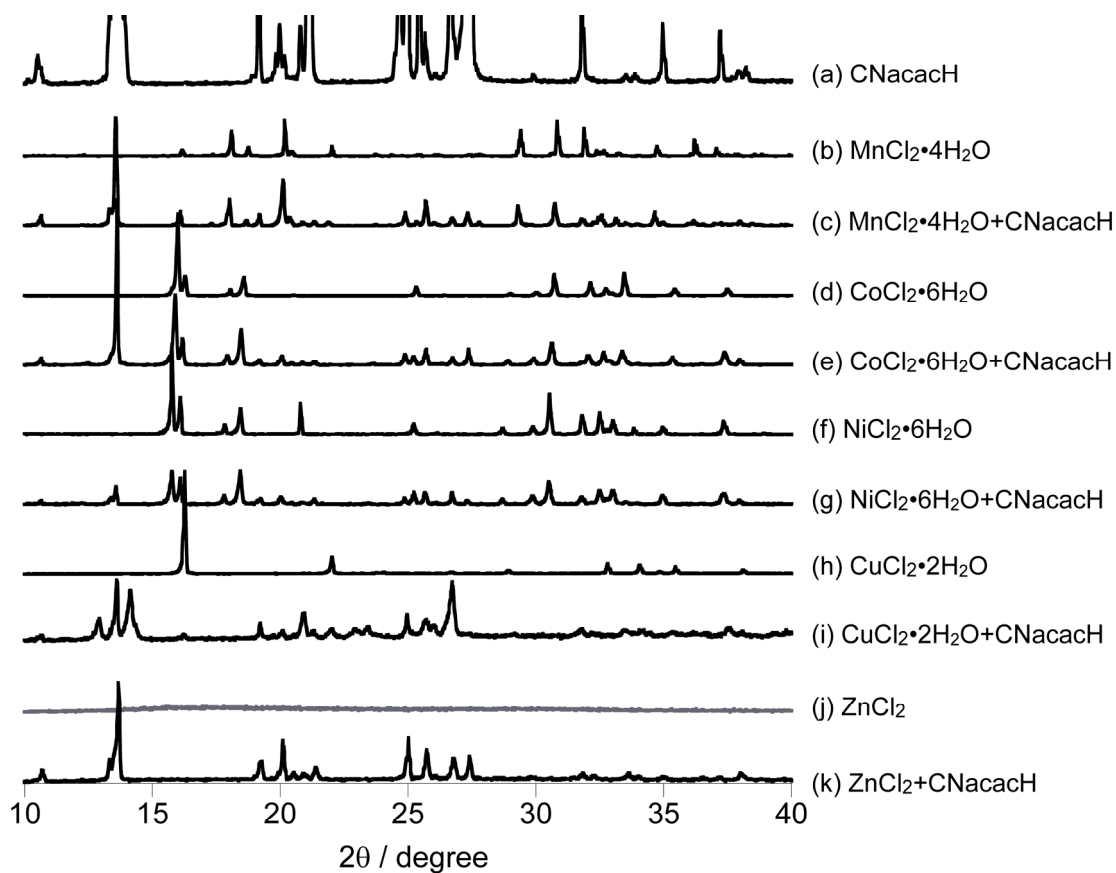


Figure S3. XPRD patterns of (a) CNacacH, (b) $\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$, (c) a mixture of $\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$ and CNacacH after 5 min grinding, (d) $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$, (e) a mixture of $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$ and CNacacH after 5 min grinding, (f) $\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$, (g) a mixture of $\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$ and CNacacH after 5 min grinding, (h) $\text{CuCl}_2 \cdot 2\text{H}_2\text{O}$, (i) a mixture of $\text{CuCl}_2 \cdot 2\text{H}_2\text{O}$ and CNacacH after 5 min grinding, (j) ZnCl_2 and (k) a mixture of ZnCl_2 and CNacacH after 5 min grinding.

Table S1. Crystallographic and experimental data for new complexes synthesized in this study.

Compound	[Mn(CNacac) ₂ (H ₂ O) ₂]	[Fe ₂ (μ-OMe) ₂ (CNacac) ₄]	[Ni(CNacac) ₂ (H ₂ O) ₂].2H ₂ O	[Cu(CNacac) ₂ (H ₂ O)]	[Zn(CNacac) ₂ (H ₂ O)]
Formula	C ₁₂ H ₁₆ N ₂ O ₆ Mn	C ₂₆ H ₃₀ N ₄ O ₁₀ Fe ₂	C ₁₂ H ₂₀ N ₂ O ₈ Ni	C ₁₂ H ₁₄ N ₂ O ₅ Cu	C ₁₂ H ₁₄ N ₂ O ₅ Zn
<i>Mr</i>	339.21	670.24	379.01	329.79	331.62
Space group	<i>C2/c</i>	<i>P</i> $\bar{1}$	<i>C2/m</i>	<i>P</i> $\bar{1}$	<i>Fdd2</i>
<i>a</i> / Å	9.1222(4)	8.016(2)	7.7703(4)	7.971(5)	12.5272(6)
<i>b</i> / Å	12.5848(8)	9.868(2)	21.2435(15)	8.227(5)	27.3148(10)
<i>c</i> / Å	14.0541(8)	10.433(4)	5.1790(3)	12.025(5)	8.4374(3)
α / °	-	84.415(14)	-	77.969(5)	-
β / °	98.495(2)	76.815(14)	101.817(1)	81.610(5)	-
γ / °	-	71.185(11)	-	66.629(5)	-
<i>V</i> / Å ³	1595.72(15)	760.3(4)	836.77(9)	706.2(7)	2887.1(2)
<i>Z</i>	4	1	2	2	8
μ (Mo K α) / mm ⁻¹	0.853	1.013	1.200	1.566	1.720
^a GOF on <i>F</i> ²	0.929	1.048	1.119	1.099	1.039
^b <i>R</i> 1 [on <i>F</i> , <i>I</i> > 2 σ (<i>I</i>)]	0.0347 (1343)	0.0310 (4545)	0.0342 (945)	0.0281 (4216)	0.0388 (2273)
^c <i>wR</i> 2 (on <i>F</i> ² , all data)	0.0849 (1805)	0.0807 (5195)	0.0919 (983)	0.0791 (4850)	0.1223 (2422)

$$^a \text{GOF} = \left\{ \frac{\sum [w(F_o^2 - F_c^2)^2]}{(n - p)} \right\}^{1/2} \quad (n; \text{ number of reflections, } p; \text{ total number of parameters refined}), \quad ^b R1 = \frac{\sum (|F_o| - |F_c|)}{\sum |F_o|}$$

$$^c wR2 = \left\{ \frac{\sum [w(F_o^2 - F_c^2)^2]}{\sum [w(F_o^2)^2]} \right\}^{1/2}$$